

CO₂-development: a vision of a carbon dioxide economy

Carbon dioxide (CO₂) emissions are largely responsible for the greenhouse effect and thus for climate change. A reduction in CO₂ emissions is therefore at the very top of the international political agenda. Trials are running in parallel to explore underground sequestration of CO₂ from power stations, thereby removing it from the atmosphere. It would at first sight seem paradoxical to wish to use energy-poor, inert CO₂ molecules. Considerable research and development efforts in recent years have led to new and innovative CO₂-recycling technologies and a vision of a CO₂ economy.

CO₂ recycling has quickly become a hot topic for the future for every large company in the chemicals and plastics sector. Wirtschaftswoche reports that even Nobel prizewinners George Olah and Joseph Stiglitz have recognized the gas as a future fuel and raw material of the chemical industry. In the last three years, the US Department of Energy and the German Ministry for Research (BMBF) have each provided some €100 million for research into new uses for CO₂. These investments are already bearing fruit. Evonik, BASF and Bayer Material Science are working hard on CO₂ polymers. Siemens and BASF demonstrated the first applications in household appliances such as fridge compartments and vacuum cleaner casings at theACHEMA fair in Frankfurt in June 2012. The automobile and aircraft industries are working on fuels that depend on neither from oil nor biomass, but are instead derived from solar and wind power – and CO₂. These are also early days for a new chemical sector: recycling – the cascade use of CO₂ as a raw material for the chemical industry. Now new chemical and electrochemical reactions must be discovered and further technologies developed (e.g. the efficient separation and purification of CO₂ from the emission flow) to turn the climate killer into a renewable resource.

Alessandra Quadrelli from Lyons University sees CO₂ as one of the most important raw materials for the chemical industry in the future. According to her calculations, innovative chemical uses of CO₂ could achieve up to 10% of the global reduction in greenhouse gases that is required.

CO₂ polymers – new options for the plastic industry

The main new CO₂ polymer is polypropylene carbonate (PPC), which was first developed 40 years ago by Inoue, but is only now coming into its own. PPC is 43% CO₂ by mass, biodegradable, shows high temperature stability, high elasticity and transparency, and a memory effect. These characteristics open up a wide range of applications for PPC, including countless uses as packing film and foams, dispersions and softeners for brittle plastics. The North American companies Novomer and Empower Materials, the Norwegian firm Norner and SK Innovation from South Korea are some of those working to develop and produce PPC. Bayer Material Science exhibited

polyurethane blocks at AICHEMA, which were made from CO₂ polyols. CO₂ replaces some of the mineral oil use. Industrial manufacturing of foams for mattresses and insulating materials for fridges and buildings is due to start in 2015.

PPC as a softener for bioplastics

Many bio-based plastics, e.g. PLA and PHA, are originally too brittle and can therefore only be used in conjunction with additives for many uses. Now a new option is available. They can cover an extended range of material characteristics through combinations of PPC with PLA or PHA. This keeps the material biodegradable and translucent, and it can be processed without any trouble using normal machinery. The vacuum cleaner casings that Bosch Siemens Household Appliances (BSH) displayed at AICHEMA are predominantly made of BASF's PPC and PHA and are intended as a substitute for the bulk plastic ABS. The first internal life-cycle analysis studies demonstrate the material's clear advantages. PPC/PLA combinations were used in fridge compartments.

Fuel from wind power, solar power and CO₂

An outside energy source is required if CO₂ is to be used as fuel. The major option here is to use surplus wind and solar power, which frequently occurs in Germany. Storage is a central concern with the expansion of renewable energy. If the surplus electricity is used to produce hydrogen (H₂) from water, this can then be converted into various fuels in conjunction with CO₂. The first reaction is that of H₂ with CO₂ to form methane (CH₄), which can then be fed into the gas network. Further chemical processes lead to methanol, petrol, diesel and kerosene. The high temperature steam electrolysis that is being optimized in the BMBF project now achieves a 70% efficiency level (electricity to hydrogen). In 2011 a consortium of businesses in Iceland began building the first commercial plant, which will produce 5 million litres of methanol per year from CO₂. That would cover 2.5% of Iceland's fuel needs.

CO₂ as growth substrate for algae and bacteria

However, the world's largest use of CO₂ takes every day right in front of our eyes. With the help of photosynthesis (and with the action of sunlight), plants convert carbon dioxide into sugar, which they then use to produce all the important bio-molecules. This can also be commercially exploited: in large-scale reactors algae are gassed with carbon dioxide from power stations and then produce biomass. Some bacteria can also use CO₂. The metabolism of these so-called acetogenic bacteria enables them to use CO₂ along with a carbon monoxide/ hydrogen mixture (synthesis gas) as a growth substrate and as a basis for producing various products such as acetone, butanol and ethanol. A joint project between RWE and biotech company Brain was able to isolate numerous strains of bacteria in power station chimneys that could serve this purpose.

Changes through molecular engineering to the bacteria can also lead to products other than the normal end products - for example the acrylic acids needed to produce PMMA (a polymer better known as plexiglass) and the biopolymer PHB. Synthetic biology methods should even allow for the production of customized bacteria in future for optimal CO₂ efficiency. Evonik in particular is working on the production of various chemicals, while the New Zealand firm LanzaTech is developing aircraft fuel and specialty chemicals based on butanol derived from CO₂ fermentation.

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